

## Poster-2-26

**Metal-insulator transition in isotope substituted Vanadium Oxide**Willem Rischau, Javier Del Valle Grande, Stefano Gariglio, and Jean-Marc Triscone*University of Geneva*

Vanadium oxide  $\text{VO}_2$  is a transition metal oxide showing a metal-to-insulator transition (MIT) close to room temperature at around 340 K [1]. The decrease in electrical conductivity is accompanied by a structural change from monoclinic to a rutile lattice. Despite its discovery in the 1960s the origin of the phase transition is still debated and scenarios based on electron-lattice correlations (Peierls transition) or electron-electron correlations (Mott transition) have been proposed.

In order to explore the effect played by phonons in this phase transition, we studied the MIT in isotope-substituted  $\text{V}^{18}\text{O}_2$ . Single crystalline  $\text{V}^{16}\text{O}_2$  and  $\text{V}^{18}\text{O}_2$  microtubes were grown by thermal oxidation through resistive heating of vanadium foil [2]. XRD analysis show that the crystal structure remains unchanged upon isotope substitution. Raman spectroscopy shows the expected phonon shift towards lower energies for  $\text{V}^{18}\text{O}_2$ . The MIT observed in electrical transport measurements shifts by around 1.5 K towards higher temperatures upon isotope substitution. A possible explanation for this shift is an increased lattice entropy in  $\text{V}^{18}\text{O}_2$  making the insulating state thermodynamically more stable.

[1] Morin, Phys. Rev. Lett. 3, 1 (1959).

[2] Zhao et al., Communications Materials 1, 28 (2020).